

Periodic Table Competition 2022-23

Name of Activity: Periodic Table Competition

Date and time of activity conducted: 24/01/2023, at 11.00am

Target Group: B.Sc. 1

Number of students took the advantage: 29

Name of Organizer: Department of Chemistry

Description of activity conducted: The department of chemistry organized "Periodic Table Competition" The program was started with preface addressed by Head, department of Chemistry Mrs. R. A. Pawar madam. Rules and regulations of Periodic Table Competition were declared by Prof. Dr. C. B. Mane sir. In competition 29 students were participated. The competition was conducted in Room No. 44. The main purpose behind such competition is to enhance the knowledge as well as interest in chemistry regarding periodicity. In this competition students were draw the periodic table, labeled group and period number as well as written elements with atomic number. First three numbers were felicitated by offering prize and certificate.

Rankers

- 1] First Prize- Ms. Sharvari Sudhir Vathare
- 2] Second Prize- Ms. Namrata Shamrao Kothari
- 3] Third Prize- Amruta Maruti Chopade

The image shows a student-drawn periodic table. It includes a grid with handwritten labels for groups (I to VIII) and periods (1 to 7). Below the grid, there are handwritten lists for 'Lanthanoid series' and 'Actinoid series'. To the right of the grid, there is a list of elements with their atomic numbers, such as H (1), He (2), Li (3), Be (4), B (5), C (6), N (7), O (8), F (9), Ne (10), Na (11), Mg (12), Al (13), Si (14), P (15), S (16), Cl (17), Ar (18), K (19), Ca (20), Sc (21), Ti (22), V (23), Cr (24), Mn (25), Fe (26), Co (27), Ni (28), Cu (29), Zn (30), Ga (31), Ge (32), As (33), Se (34), Br (35), Kr (36), Rb (37), Sr (38), Y (39), Zr (40), Nb (41), Mo (42), Tc (43), Ru (44), Rh (45), Pd (46), Ag (47), Cd (48), In (49), Sn (50), Sb (51), Te (52), I (53), Xe (54), Ba (56), La (57), Ce (58), Pr (59), Nd (60), Pm (61), Sm (62), Eu (63), Gd (64), Tb (65), Dy (66), Ho (67), Er (68), Tm (69), Yb (70), Lu (71), Hf (72), Ta (73), W (74), Re (75), Os (76), Ir (77), Pt (78), Au (79), Hg (80), Tl (81), Pb (82), Bi (83), Po (84), At (85), Rn (86), Fr (87), Ra (88), Ac (89), Th (90), Pa (91), U (92), Np (93), Pu (94), Am (95), Cm (96), Bk (97), Cf (98), Es (99), Fm (100).

Co-ordinator: Mrs. R. A. Pawar

Head
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